

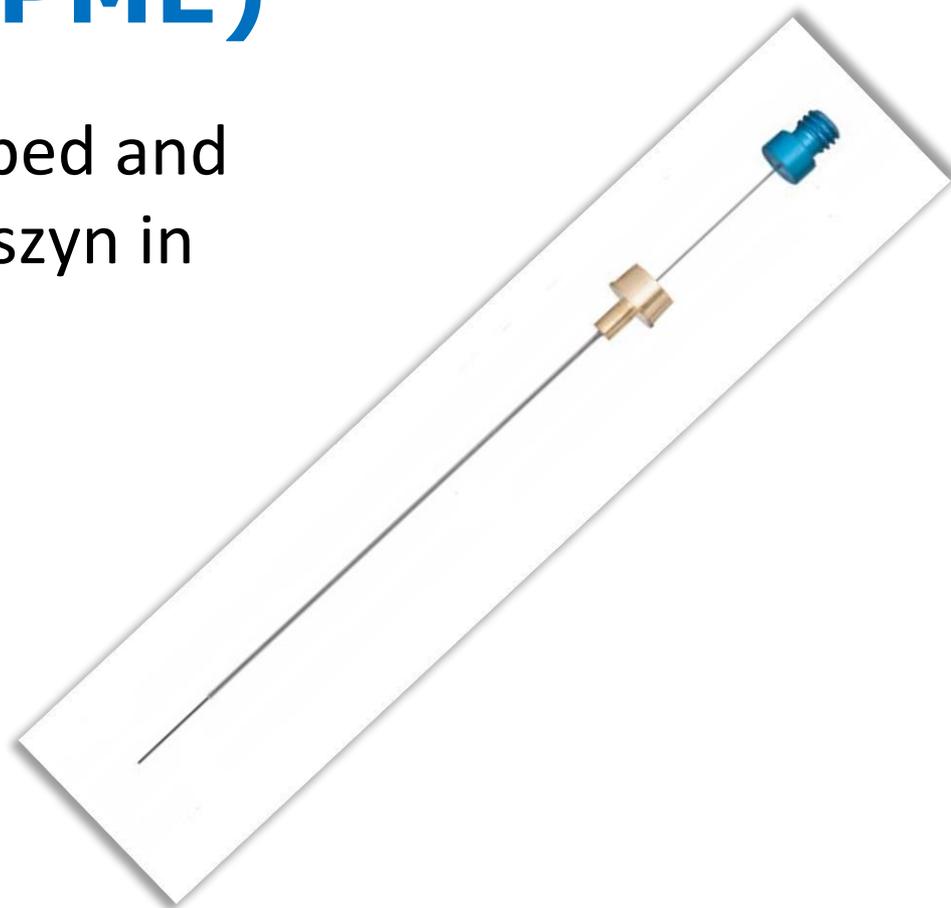
Evaluation and Application of SPME Arrows

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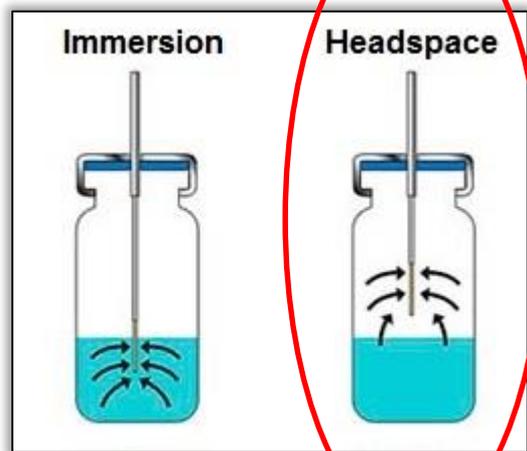
Solid Phase Microextraction (SPME)

- SPME fibers were developed and patented by Janusz Pawliszyn in 1990
- Subsequently licensed to Supelco until 2014

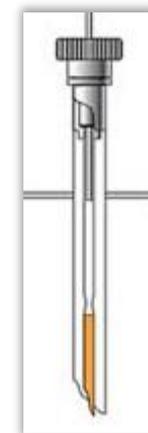


SPME Sampling/Desorbing

- Target analytes are collected via headspace or immersion
- Desorbed in the GC inlet

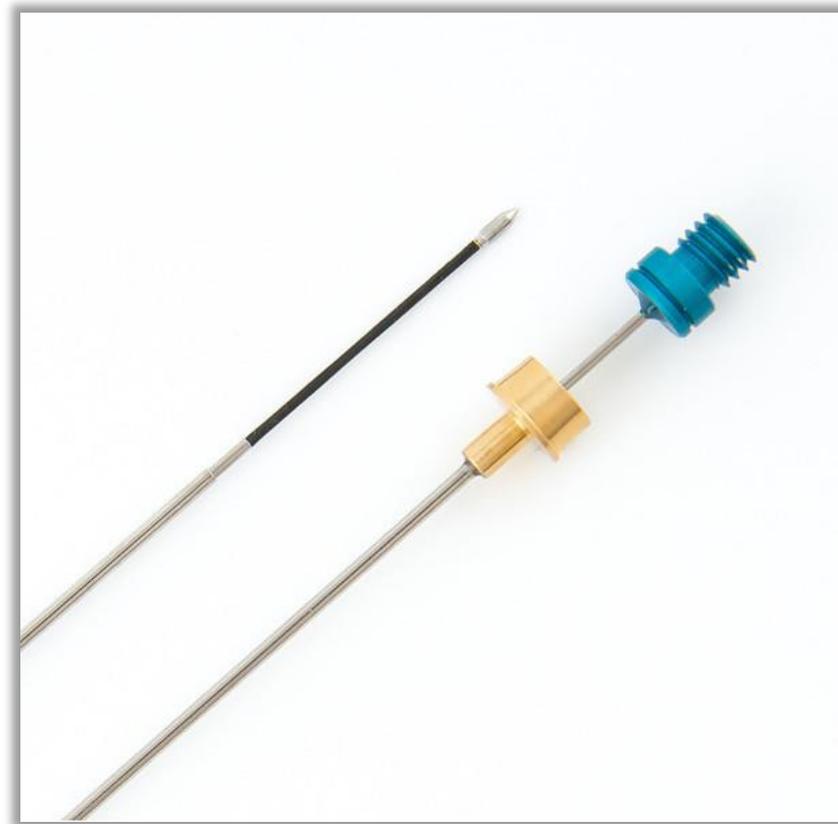


90% of applications



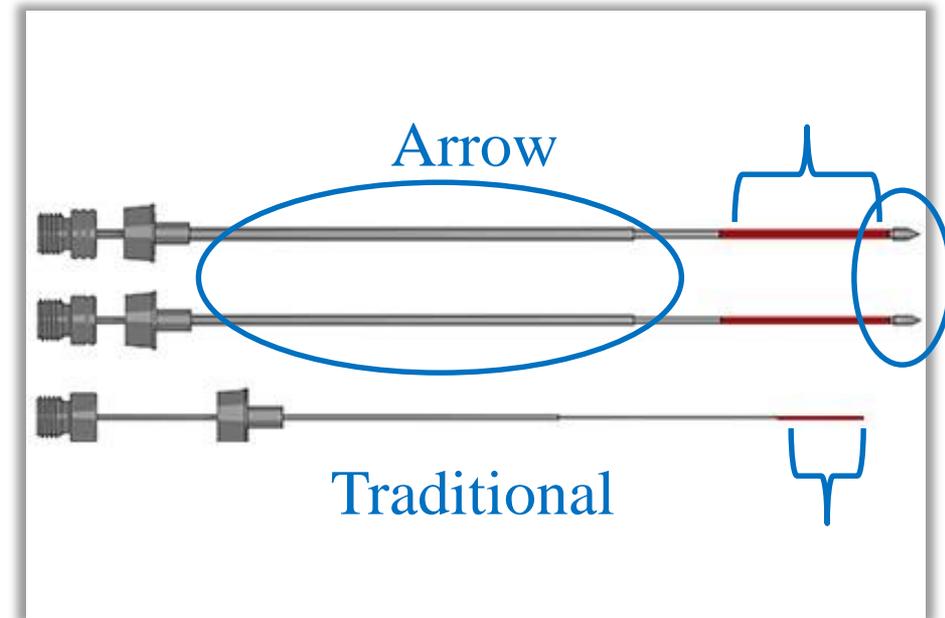
SPME Arrow

- Developed by CTC Analytics AG
 - PAL Systems (“Rails”)
- To overcome the limited mechanical stability and small phase volumes



Traditional vs Arrow

- Arrow tip facilitates smooth septa penetration
- Fully protects phase
- Thicker needle is more durable
- Larger phase area/volume for increased sensitivity



Phase Volumes

- Initial design of up to 6x the area and 20x the volume
- Our calculations show 17 - 18x the volume
- Sensitivity (up to 10x) increase seems intuitive and reasonable



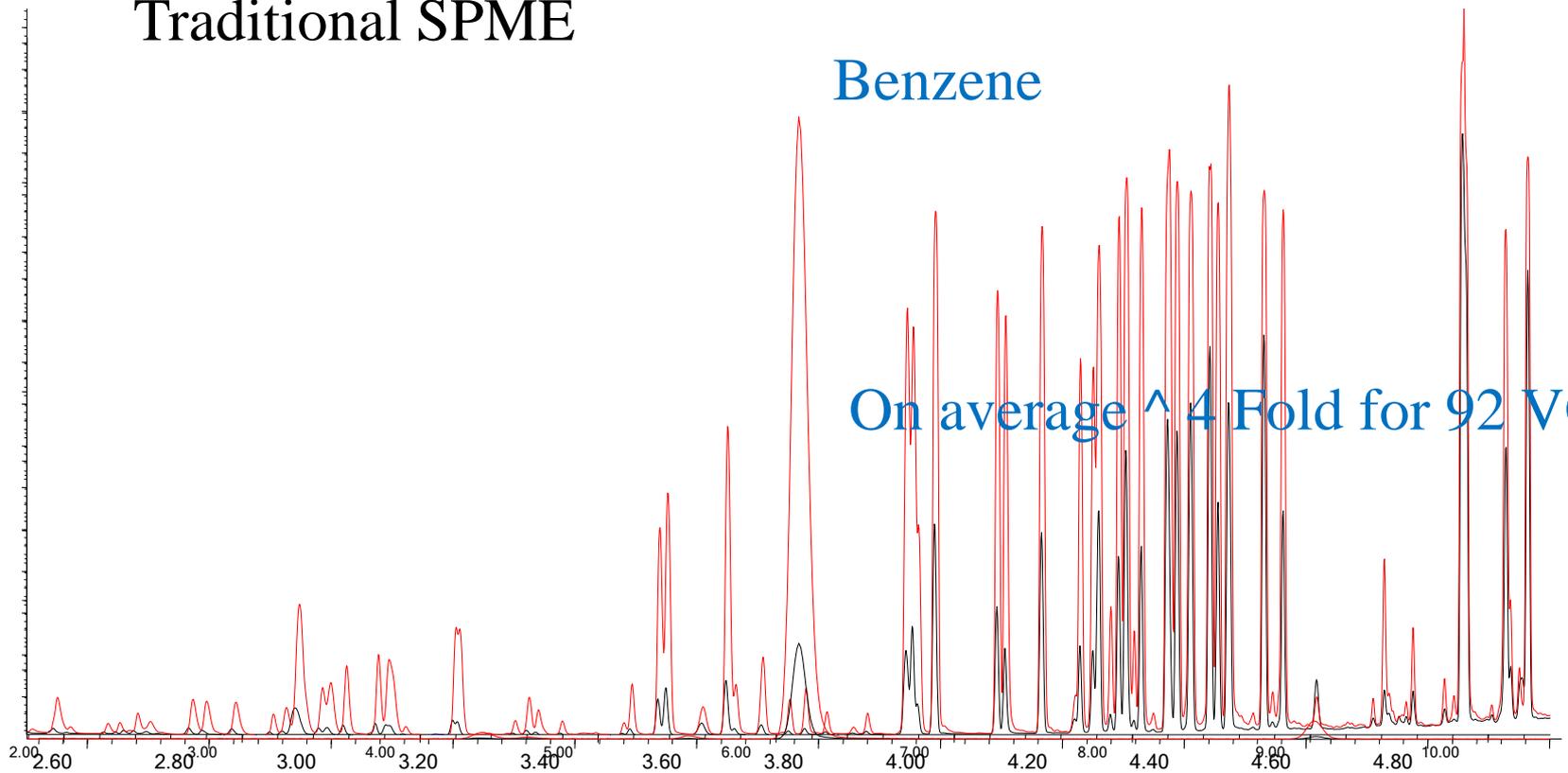
Sensitivity of HS VOCs in Water

- 10 mL of H₂O
- 2.5 ppb of ISO 17943 VOCs
- 3 g NaCl
- 2 min of equilibration @ 60 °C
- 100 µm PDMS traditional and Arrow
- 10 min of headspace extraction @ 60 °C w/ 500 rpm of heatex

VOCs in Water

SPME Arrow (1.1 mm)

Traditional SPME

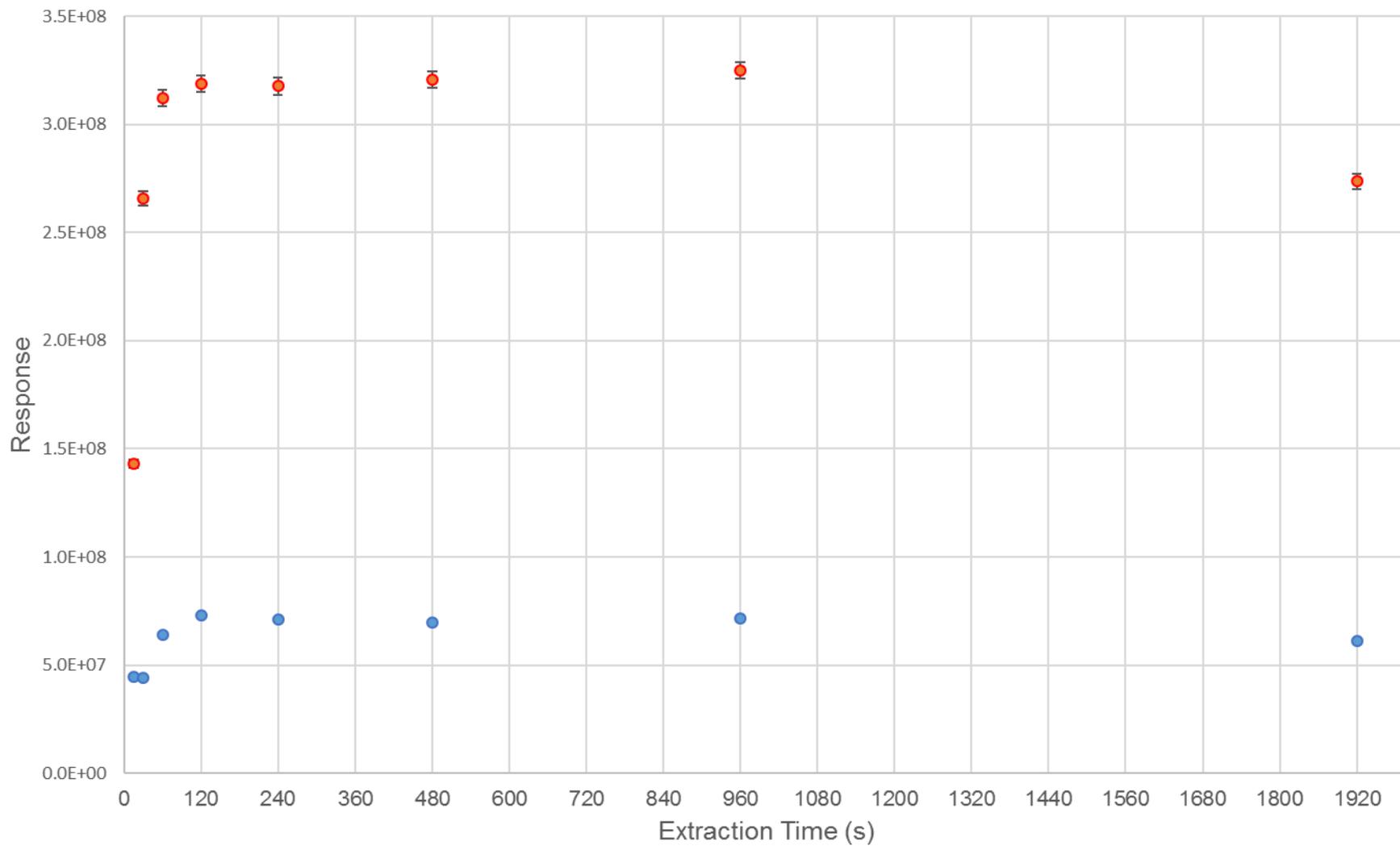


Response vs Extraction Time

- Goal: utilize the Arrow with shorter extraction times to achieve the same sensitivity as traditional SPME
- Evaluated traditional and Arrow for:
 - 15, 30, 60, 120, 240, 480, 960, and 1920 s

Sum of Responses @ 2.5 ppb vs Extraction Time

—●— Traditional —●— Arrow



Ruggedness / Lifetime

- Rack up a traditional fiber and Arrow
- Run 24 to 48 samples / day
 - Same # of runs on each SPME
- Evaluate responses everyday
- First to 50% collection efficiency or a mechanical failure loses

A far too common outcome...



Real Ruggedness Example



SPME Methods

- **ASTM D 6520, 2000**
 - Standard Practice for the SPME of Water and its Headspace for the Analysis of Volatile and Semi-Volatile Organic Compounds
- **ASTM D 6889, 2003**
 - Standard Practice for Fast Screening for Volatile Organic Compounds in Water Using Solid Phase Microextraction (SPME)
- **EPA Method 8272, 2007**
 - Parent and Alkyl PAHs in Sediment Pore Water by SPME-GC/MS
- **ISO 27108 (DIN 38407-34), 2013**
 - Determination of selected plant treatment agents and biocide products - Method using SPME followed by GC-MS
- **ISO 17943 (DIN 38407-41)**
 - Determination of VOCs in water - GC-MS after HS-SPME

ISO 17943

- We evaluated 92 VOCs. ISO 17943 only list 63 VOCs.
- Calibration linearity, precision, and detection limits were obtained from:
 - 10 mL of drinking water
 - With only a 2 minute headspace extraction @ 60 °C
 - With a 1 minute desorption @ 250 °C

Compound	%RSD		
	(0.0025 - (n = 7 @ 250 166 µg/L) ng/L)	(ng/L)	
Dichlorodifluoromethane (CFC-12)	0.992	3.16	8.73
Chloromethane	0.997	12.3	135
Vinyl chloride	0.997	7.51	24.2
Bromomethane	0.999	18.3	173
Chloroethane	0.994	30.1	102
Trichlorofluoromethane (CFC-11)	0.991	1.97	2.75
Diethyl ether	0.998	5.11	11.1
1,1-Dichloroethene	0.998	3.09	7.25
Carbon disulfide	0.997	2.11	8.52
Trichlorotrifluoroethane (CFC-113)	0.992	6.73	8.08
Iodomethane	0.997	3.29	5.61
Allyl chloride	0.998	1.53	3.17
Methylene chloride	0.998	2.92	54.0
Acetone	0.996	5.76	137
trans-1,2-Dichloroethene	0.999	1.91	7.78
Methyl tert-butyl ether (MTBE)	0.999	5.51	11.2
tert-Butanol (TBA)	0.999	6.97	28.5
Diisopropyl ether (DIPE)	0.999	14.1	39.7
Acetonitrile	0.999	3.01	169
Chloroprene (2-chloro-1,3-Butadiene)	0.999	3.18	17.4
1,1-Dichloroethane	0.999	3.17	7.03
Acrylonitrile	0.999	3.18	17.4
Ethyl-tert-butyl ether (ETBE)	0.999	4.36	9.34
cis-1,2-Dichloroethane	1.000	2.49	8.69
2,2-Dichloropropane	0.998	3.07	53.0
Bromochloromethane	0.999	2.51	5.69
Chloroform	1.000	1.00	22.6
Methyl acrylate	0.989	3.42	47.1
Carbon tetrachloride	1.000	13.3	29.2
Tetrahydrofuran	0.978	0.32	1456
1,1,1-Trichloroethane	1.000	2.66	6.74
2-Butanone (MEK)	0.954	4.86	2843
1,1-Dichloropropene	1.000	2.21	10.8
Benzene	0.999	3.20	31.3
Propionitrile	0.989	4.64	89.6
Methacrylonitrile	1.000	7.04	33.1
tert-Amyl methyl ether (TAME)	0.998	4.50	13.6
1,2-dichloroethane	1.000	2.87	10.6
Isobutyl alcohol	0.992	5.77	1702
Trichloroethene	1.000	1.37	16.3
Dibromomethane	0.996	2.25	5.3
1,2-Dichloropropane	0.997	4.96	15.9
Bromodichloromethane	0.999	2.33	11.6
Methyl methacrylate	0.999	6.24	30.1
1,4-Dioxane	0.987	3.76	396
2-Chlorethyl vinyl ether	0.999	4.16	16.2
2-Chloropropanol	0.999	6.83	407
cis-1,3-Dichloropropene	1.000	1.93	11.2
Toluene	0.995	1.88	83.6
2-Nitropropane	0.990	4.26	101
2-Hexanone	1.000	4.66	26.9
trans-1,3-Dichloropropene	0.990	2.34	34.0
Tetrachloroethene	0.997	2.74	49.5
Ethyl methacrylate	0.994	3.69	39.7
1,1,2-Trichloroethane	0.999	3.59	14.6
Dibromochloromethane	1.000	3.14	14.0
1,3-Dichloropropane	1.000	3.54	16.0
1,2-Dibromomethane	1.000	3.35	15.3
4-Methyl-2-pentanone (MIBK)	0.999	5.44	29.8
Ethylbenzene	0.989	2.34	115
Chlorobenzene	0.999	2.37	24.7
1,1,1,2-Tetrachloroethane	1.000	4.59	43.0
m-Xylene	0.981	2.47	50.8
p-Xylene	0.981	2.47	50.8
o-Xylene	0.989	2.93	38.1
Styrene	0.999	3.14	31.0
Bromoform	0.993	4.18	26.9
Isopropylbenzene	0.989	3.87	56.3
cis-1,4-Dichloro-2-butene	0.999	6.22	70.7
n-Propylbenzene	0.985	2.48	42.4
1,1,2,2-Tetrachloroethane	0.997	4.10	31.0
Bromobenzene	0.996	2.17	26.2
1,3,5-Trimethylbenzene	0.986	2.74	41.5
2-Chlorotoluene	0.996	2.11	24.9
trans-1,4-Dichloro-2-butene	0.997	2.79	30.2
sec-Butylbenzene (1-methylpropyl)	1.000	3.00	30.6
4-Chlorotoluene	0.996	2.11	24.9
tert-Butylbenzene	0.990	2.69	34.9
1,2,4-Trimethylbenzene	0.987	2.19	31.8
Pentachloroethane	0.998	5.14	41.1
sec-Butylbenzene (1-methylpropyl)	0.987	3.30	53.7
p-Isopropyltoluene	0.990	3.99	47.7
1,3-Dichlorobenzene	1.000	2.33	22.8
1,4-Dichlorobenzene	1.000	1.70	16.4
n-Butylbenzene	0.995	1.95	11.0
1,2-Dichlorobenzene	1.000	3.81	33.6
1,2-Dibromo-3-chloropropane	0.985	5.50	31.1
Hexachlorobutadiene	0.999	3.30	30.0
Nitrobenzene	0.971	28.9	198
1,2,4-Trichlorobenzene	0.999	1.27	10.4
Naphthalene	0.998	3.39	33.0
1,2,3-Trichlorobenzene	0.998	2.80	21.6
Median	0.998	3.24	30.0

ISO 17943

- Calibration
 - 3 nines or better for most compounds at 2.5 ppt to 150 ppb
- Precision
 - 3 % RSDs, which are well below method requirements
- MDLs
 - ~30 ng/L
- All from a 2 minute extraction!!!

Implications

- SPME Arrow is more sensitive and sturdy than traditional SPME fibers
- Demonstrated excellent linearity, precision, and detection limits
- However, limited method applicability in the U.S.

SPME Arrow Application

- Fast screening!!!
 - ASTM D 6889, 2003
 - Standard Practice for [Fast Screening](#) for Volatile Organic Compounds in Water Using Solid Phase Microextraction (SPME)
- A 2 minute extraction and 4 minute GC-MS screening run could prevent hours of purge-and-trap instrument down time from a hot sample

Thank You!

- Colton Myers
- Co-authors
- CTC Analytics AG